#### **CHAPTER THREE**

#### 3.0 MATERIALS AND METHODS

## 3.1 General Experimental Procedure, Solvents and Fine Chemicals

Melting points were determined using Gallenkamp melting point apparatus (Manchester, UK) and are uncorrected. Optical rotation was measured on a Jasco P-1020 Polarimeter (Jasco Corporation, Tokyo, Japan). UV spectra were analysed using a Shimadzu UV-2401 A spectrophotometer (Shimadzu corporation, Kyoto, Japan). IR data were recorded on a Bruker Tensor 27 FTIR spectrophotometer (Bruker Corporation, Bremen, Germany) as KBr pellets. NMR data were measured in CDCl<sub>3</sub>, CD<sub>3</sub>OD and DMSO-d<sub>6</sub> on a JOEL NMR instrument operating 600 and 150 MHz, respectively. Some NMR data were done using Brucker AM 300 spectrometer operating at 400, 300 and 125, 75 MHz, respectively. Chemical shifts are expressed in ppm with tetramethylsilane (TMS) used as internal standard. The mass spectral data were obtained using a Varian MAT 8200 A instrument. Column chromatography was performed using silica gel 60 (0.063 - 0.200 mm, Merck-Germany) while thin layer chromatography (TLC) was performed using silica gel 60 Å F<sub>254</sub> (Merck) pre-coated plates. Paper chromatography was done on standard Whatman No. 1 chromatography paper. Shift reagents were prepared according to Mabry *et al* (1970). All solvents used were of analytical grade.

#### 3.2 Plant materials, collection and identification

Lonchocarpus eriocalyx (Harms) (Reference No.: LE/58/2013) was collected from Embu-Mbeere (Lat: 0.5833° S and Long: 37.6333° E) while *Alysicarpus ovalifolius* (Schumach) (Reference No.: FAB/AO/2012) was obtained from Shimba Hills (Lat: 4° 19' 39" S and Long: 39° 21' 39" E) where the plants grow naturally. The leaves of *Erythrina abyssinica* (DC) (Reference No.: MU/EA/76/2013) were collected from Rawalo Hills (Lat: 5° and Long: 34° 30'

E) which is in Central Gem, Siaya County. The plant materials were authenticated at the herbarium of the National Museums of Kenya where voucher specimens are preserved.

#### 3.3 Extraction of plant materials

#### 3.3.1 Extraction of the stem bark of *Lonchocarpus eriocalyx*

The air dried and pulverized stem bark (2 kg) was soaked sequentially in *n*-hexane (3 x 3L), CH<sub>2</sub>Cl<sub>2</sub> (3 x 3L) and MeOH (3 x 3L), each lasting four days at room temperature. The extracts were separately filtered and evaporated under reduced pressure to afford yellow (5 g), yellowish-brown (25 g) and reddish-brown (106 g) extracts, respectively.

## 3.3.2 Thin layer chromatography (TLC) analysis of n-hexane and $CH_2Cl_2$ extracts

TLC analysis of n-hexane extract revealed two components with  $R_f$  values 0.82 and 0.63 [n-hexane-CH<sub>2</sub>Cl<sub>2</sub> (2:3)] which turned greenish-purple with anisaldehyde spraying reagent. On the other hand, TLC analysis of CH<sub>2</sub>Cl<sub>2</sub> extract using n-hexane-CH<sub>2</sub>Cl<sub>2</sub> (2:3) afforded five spots of  $R_f$  values 0.82, 0.63, 0.48, 0.32 and 0.18 which turned bluish-purple using anisaldehyde spraying reagent.

#### **3.3.3** Fractionation of *n*-hexane extract

A portion of the *n*-hexane extract (4 g) was mixed with 4 g of silica gel in 20 mL of CH<sub>2</sub>Cl<sub>2</sub>. The solvent was removed using a rotary evaporator and the free flowing solid chromatographed over silica gel packed column (2.0 x 60 cm, 120 g) using *n*-hexane with increasing amount of CH<sub>2</sub>Cl<sub>2</sub> up to 100% of the latter. This afforded 60 fractions, each 20 mL which were collected and their homogeneity monitored by TLC (solvent systems: *n*-hexane EtOAc, 9:1 and 4:1). The eluants were grouped into three major pools (**I-III**) depending on TLC profiles. Fractions 1-15 constituted pool **I**, which upon evaporation of the solvent afforded a yellow oily material that lost color with time and was not followed further. Pool **II** (fractions 20-40, 1.5 g) showed a single

spot of  $R_f$  0.82 (solvent system: *n*-hexane-EtOAc, 4:1) which on further recrystallization gave **133** (55 mg) as white amorphous powder. Pool **III** (fractions 41-57) showed a single spot of  $R_f$  0.63 (eluant, *n*-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 2:3) which crystallized in *n*-hexane CH<sub>2</sub>Cl<sub>2</sub> mixture to give compound **134** as white needle-like crystals (100 mg).

#### 3.3.4 Physical and spectroscopic data of compounds from *n*-hexane extract

#### **3.3.4.1** Compound 133

White needle-like crystals,  $R_f$ , 0.82 [n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, (2:3)], m.p. 254-256°C (Lit. 260-262°C; Majidul et~al., 2015);  ${}^{1}$ H and  ${}^{13}$ C NMR data: (See **Table 10, page 73, Appendix 4**); **EI-MS**: m/z (rel int.): 426 [ $M^{+}$ ] (25), 408 (10), 343 (18), 331 (20), 316 (9), 154 (10), 127 (17), 97 (30), 73 (100).

#### **3.3.4.2** Compound 134

White needle-like crystals  $R_f$ , 0.63 [n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, (2:3)], m.p. 132-134°C (Lit. 136-138°C; Orabi, 2011); IR  $v_{max}$ , (KBr) cm<sup>-1</sup>: 3470.3 (OH), 2933. 2859.3 1693.5 (C=C), 1457.0, 1270.4, 995.7, 926.4. H and  $^{13}$ C NMR data: (See **Table 11**, **page 76**, **Appendix 5**); **EI-MS**: m/z (rel int.): 414 [M]  $^+$  (100), 396 (70), 381 (40), 329 (50), 303 (50).

### 3.3.5 Fractionation of CH<sub>2</sub>Cl<sub>2</sub> extract

A portion of the CH<sub>2</sub>Cl<sub>2</sub> extract (20 g) was adsorbed onto 20 g silica gel and then subjected to column chromatography (3.0 x 60 cm, SiO<sub>2</sub>, 200 g, and pressure≈1 bar) using *n*-hexane, *n*-hexane-CH<sub>2</sub>Cl<sub>2</sub> gradient (increment 10%) up to 100% CH<sub>2</sub>Cl<sub>2</sub>) and elution concluded with 100% ethyl acetate. A total of 200 fractions, each 20 ml were collected. This process afforded subfractions (I-IV) as determined by TLC profiles [solvent systems: *n*-hexane-CH<sub>2</sub>Cl<sub>2</sub> (4:1, 1:3, 2:3) and CH<sub>2</sub>Cl<sub>2</sub>-MeOH (99:1)]. Sub-fraction I (fractions 1-30) mainly eluted with *n*-hexane did not show clear spots on TLC and was not followed further. Sub-fraction II (fractions 35-90) afforded

yellow oil which became colourless and was not follower further. Sub-fraction **III** (fractions 92-141) showed two spots of  $R_f$  values of 0.82 and 0.63 (eluant: n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 2:3) which on evaporation of the solvent followed by recrystallization in n-hexane-CH<sub>2</sub>Cl<sub>2</sub> mixture afforded a further amount of **133** ( $R_f$ , 0.82, 45 mg). The mother liquor of this sub-fraction was further purified using medium pressure liquid chromatography eluting with n-hexane-CH<sub>2</sub>Cl<sub>2</sub> (4:1) as the mobile phase to give a further amounts of **133** (15 mg) and compound **135** ( $R_f$ , 0.63, 40 mg). Lastly, sub-fraction **IV** (fractions 143-200), 6 g) upon repeated chromatrographic separation afforded a further amounts of **134** ( $R_f$ , 0.63, 15 mg), **135** as white powder ( $R_f$ , 0.48, 40 mg) and **136** as amorphous white powder ( $R_f$ , 0.32, 70 mg) and **27** ( $R_f$ , 0.18, 300 mg) [TLC solvent system; n-hexane-CH<sub>2</sub>Cl<sub>2</sub> (2:3)].

## 3.3.6 Physical and Spectroscopic data of compounds from CH<sub>2</sub>Cl<sub>2</sub> extract

#### **3.3.6.1** Compound 27

White needle-like crystals,  $R_f$ , 0.18 [*n*-hexane-  $CH_2Cl_2$ , (2:3)], m.p. 214-216°C (Lit. 215-216°C; Saratha *et al.*, 2011); IR  $v_{max}$  (KBr) cm<sup>-1</sup>: 3315, 2900, 1650, 1462, 1190, 1037, 997, 681; <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 7**, **page 66**, **Appendix 1**); **EIMS**: m/z (rel int.): 426 [M] <sup>+</sup> (12), 238 (60), 180 (22).

#### 3.3.6.2 Compound 135

White powder,  $R_f$ , 0.48 [n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, (2:3)], m.p. 254-256°C (Lit. 260-262°C, Aher et~al., 2010)  $^1$ H and  $^{13}$ C NMR data  $\delta$  ppm: (See **Table 12, page 79**, **Appendix 6**); **EI-MS**: m/z (rel int.): 424 [M]  $^+$  (30), 318 (27), 206 (75), 189 (35), 109 (60).

#### 3.3.6.3 Compound 136

Amorphous white powder  $R_f$ , 0.32 [n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, (2:3)], m.p. 270-272 °C (Lit. 272-274 °C; Mahbuba  $et\ al.$ , 2012); <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 13**, **page 82**, **Appendix 7**); **EI-MS**: m/z (rel int): 576 [M+2]<sup>+</sup> (7), 574 [M] <sup>+</sup> (6), 559 (1.5), 531.

#### 3.3.7 Fractionation of MeOH extract

A portion of the powdered extract (100 g) was mixed with 20 g of silica gel and put in a desiccator with a drying agent to remove any traces of water. The sample was then subjected to column chromatography on oxalic acid deactivated silica gel (5.0 x 60 cm, 500g, pressure ≈1 bar), starting with CH<sub>2</sub>Cl<sub>2</sub> followed by CH<sub>2</sub>Cl<sub>2</sub>/MeOH gradient with increasing concentration of the polar solvent (increment 3%) and elution concluded until 100% MeOH was used. A total of 300 fractions each 20 mL were collected and their homogeneity determined by TLC (eluent: CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 99:1, 98:2, 97:3, 95:5, 4:1 and 4:1; n-BuOH/HOAc/H<sub>2</sub>O, 4:5:1) and those exhibiting similar profiles were combined into five major pools (1-VI). Pool I (fractions 1-30, 4 g) eluted using CH<sub>2</sub>Cl<sub>2</sub> afforded a mixture of compounds with R<sub>f</sub> values of 0.63, 0.18 and 0.32 and were resolved into individual components using medium pressure chromatography as already described in subsections 3.3.5 to give a further amount of 27 (15 mg), 134 (20 mg) and 135 (30 mg). Fractions 32-80 (6 g) constituted pool II and was similarly chromatographed as described above using CH<sub>2</sub>Cl<sub>2</sub>-MeOH (99:1) followed by the same solvent system in the ratio 98:2 to give 137 (R<sub>f</sub> 0.42, 30 mg) and 68 (R<sub>f</sub> 0.39, 31 mg, (CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 98:2). Pool III (fractions 83-108) showed one major spot of R<sub>f</sub> value of 0.39 (solvent system CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 98:2) and was purified by crystallization from MeOH-H<sub>2</sub>O mixture and yielded more amount of 68 ( $R_f$  0.39, 33 mg). Pool IV (fractions (110-153, 11 g) showed two spots of  $R_f$  values 0.5 and 0.36 (CH<sub>2</sub>Cl<sub>2</sub>-MeOH (97:3) and were separated using the same solvent system into compounds

**65** (R<sub>f</sub> 0.5, 90 mg) and **138** (R<sub>f</sub> 0.36, 25 mg). Fractions 157-230 (10 g) constituted pool **V** which upon repeated medium pressure chromatographic separation using CH<sub>2</sub>Cl<sub>2</sub>-MeOH (97:3) followed by the same solvent system in the ratio 96:4 and 96:4 gave a further amount of **65** (R<sub>f</sub> 0.5, 15 mg), **139** (R<sub>f</sub> 0.21, 25 mg) and **140** (R<sub>f</sub> 0.17, 45 mg). Pool **VI** (5.5 g) similarly on repeated flash chromatography using CH<sub>2</sub>Cl<sub>2</sub>-MeOH (96:4 and 95:5) afforded two compounds which were crystallized from MeOH-H<sub>2</sub>O mixture to give a further amount of **139** (R<sub>f</sub> 0.21, 68 mg) and **141** (R<sub>f</sub> 0.30, 45 mg).

## 3.3.8 Physical and spectroscopic data for compounds isolated from MeOH extract of *Lonchocarpus eriocalyx*

#### **3.3.8.1** Compound 65

Pale yellow amorphous powder  $R_f$  0.5 [CH<sub>2</sub>Cl<sub>2</sub>-MeOH, (98:2)], m.p. 314-316°C (Lit. 316-318°C; Esra, *et al.*, 2015); UV  $\lambda_{max}$ , (MeOH): 358, 298, 258 nm; IR  $\nu_{max}$  (KBr): 3500-2500 (OH), 1610 (conjugated C=O), 1450, 1340, 1250, 930 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 8**, **page 68**, **Appendix 2**); ESI-MS: m/z (rel int.): 302 [M]<sup>+</sup> (100), 274 (08), 153 (11), 137 (20), 69 (10).

#### 3.3.8.2 Compound 68

Yellow amorphous powder,  $R_f$  0.39 [CH<sub>2</sub>Cl<sub>2</sub>-MeOH, (98:2)], m.p. 346-348°C (Lit. 345-350°C; Chaturvedula and Prakash, 2013); UV  $\lambda_{max}$ , (MeOH): 268 and 337 nm; <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 9**, **page 71**, **Appendix 3**); ESI-MS: m/z (rel int.) 270 [M+1]<sup>+</sup> (100), 178 (10), 153 (15), 121 (5).

#### **3.3.8.3** Compound 137

Yellow powder  $R_f$  0.42 [CH<sub>2</sub>Cl<sub>2</sub>-MeOH, (98:2)], m.p. 291-292°C (Lit. 285-286°C; Miyaichi *et al.*, 2006]; <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 14** and **15**, **page 85**, **Appendix 8**); **ESI-MS**: m/z (rel int.): 255, [M+1]  $^+$  (62), 210 (47), 186 (32), 145(75), 110 (52), 104 (100).

## **3.3.8.4** Compound 138

Light yellow powder,  $R_f$  0.36 [CH<sub>2</sub>Cl<sub>2</sub>-MeOH, (98:2)], m.p. 301-303°C (Lit. 300-301°C; Liu *et al.*, 2012; Younghee, 2012); UV  $\lambda_{max}$  (MeOH): 254, 324 and 370 nm; IR  $\nu_{max}$  (KBr) <sup>1</sup>: 3400 (OH), 1650 (α,β-unsaturated C=O), 1600, 1450 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 14** and **15**, **page 85**, **Appendix 9**); **ESI-MS**: m/z (rel int.) 332 (20), 289 (100), 233(70), 177 (36), 109 (17).

## **3.3.8.5** Compound 139

Yellow powder, R<sub>f</sub> 0.21 [CH<sub>2</sub>Cl<sub>2</sub>-MeOH, (96:4)], m.p. 227-230°C (Lit. 225-227°C; Ahmad *et al.*, 2012; Ana *et al.*, 2009); <sup>1</sup>H and <sup>13</sup>C NMR (400 and 100 MHz, DMSO-d<sub>6</sub>) data: (See **Table 16, page 88, Appendix 10**); **ESI-MS**: *m/z* (rel int.): 463 [M-1] <sup>+</sup> (20), 423 (7), 306 (4), 342 (7), 301 (100).

## **3.3.8.6** Compound 140

Red amorphous powder,  $R_f$  [CH<sub>2</sub>Cl<sub>2</sub>-MeOH (95:5)], 0.17, m.p. 220-222°C (Lit. 223-226°C; Liu *et al.*, 2015); <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 17, page 90, Appendix 11**); **ESI-MS**: m/z (rel int.): 390 [M]  $^+$  (100).

## **3.3.8.7 Compound 141**

Greenish-yellow crystalline solid  $R_f$  0.3 [CH<sub>2</sub>Cl<sub>2</sub>-MeOH (95:5)], m.p. 234-236 °C (Lit. 238-240 °C; Okoth, 2013; Okoth *et al.*, 2013); <sup>1</sup>H NMR and <sup>13</sup>C NMR data data: (**See Table 18, page 93, Appendix 12**); ESI-MS: m/z (rel int.): 633 [M +Na] <sup>+</sup> (70), 611 (30), 465 (25), 273 (10), 215 (20).

# 3.3.9 Extraction, isolation and identification of the pure isolates from the root bark of *Alysicarpus ovalifolius*

#### 3.4.0 TLC analysis of *n*-hexane of extract

Only the root bark of *A. ovalifolius* showed positive results upon preliminary bioactivity analysis hence the aerial part was not investigated. The air dried and pulverized root bark (1 kg) of the plant was soaked sequentially in *n*-hexane (3 x 3 L), CH<sub>2</sub>Cl<sub>2</sub> (3 x 3L) and MeOH (3 x 3 L), each lasting four days at ambient temperature. The extracts were separately filtered and evaporated under reduced pressure to obtain yellow (7 g), brown (25 g) and reddish-brown (176 g) extracts respectively.

TLC analysis of the extract using n-hexane-CH<sub>2</sub>Cl<sub>2</sub> (2:3) showed two spots of  $R_f$  values 0.63 and 0.67. The spot with  $R_f$  value of 0.63 appeared reddish-purple upon spraying with anisaldehyde reagent and heating while the second spot gave purple colour on exposure to concentrated ammonia suggesting the presence of a sterol/and/or a terpenoid and a quinone derivatives, respectively.

#### **3.4.1** Fractionation of the *n*-hexane extract

The n-hexane extract (5 g) was adsorbed onto same amount of silica in 30 mL of CH<sub>2</sub>Cl<sub>2</sub>, solvent removed under reduced pressure and chromatographed over silica gel packed column (2.0 x 60 cm, 120 g) using n-hexane containing increasing amounts of CH<sub>2</sub>Cl<sub>2</sub> (up to 100%). A total of 120 fractions, each 20 mL were collected and their homogeneity monitored by TLC (solvent systems: n-hexane-EtOAc, 9:1, 4:1 and 2:3). The eluants were grouped into three pools (**I-III**) depending on TLC profiles. Fractions 1-25 constituted pool **I**, which upon evaporation of solvent afforded a yellow oily compound that lost colour with time and was not followed further. Pool **II** (fractions 30-55, 1.5 g) showed a single yellow spot of  $R_f$  0.63 (solvent system: n-hexane-

EtOAc, 4:1) which on recrystallization gave **134** (90 mg). Pool **III** (fractions 57-85, 1.0 g) crystallized [*n*-hexane-EtOAc, 4:1] to give an orange solution which on further re-crystallization afforded **142** (R<sub>f</sub>, 0.34, 75 mg).

#### 3.4.2 Physical and spectroscopic data of compounds from *n*-hexane extract

## **3.4.2.1 Compound 142**

An orange amorphous powder,  $R_f$  0.34 [n-hexane-EtOAc (4:1)], m.p. 74-76°C (Lit. 76-78°C; Sing et al., 2012; Tangmouo et al., 2005); <sup>1</sup>H NMR and <sup>13</sup>C NMR data: (See **Table 19** on **page 95**, **Appendix 13**); **ESI-MS**: m/z (rel int.) 188 [M]<sup>+</sup> (100).

## 3.4.3 Fractionation of the CH<sub>2</sub>Cl<sub>2</sub> extract

The CH<sub>2</sub>Cl<sub>2</sub> extract (22 g) was adsorbed onto same amount of silica gel and then subjected to column chromatography (3.0 x 60 cm, 240 g, pressure≈1 bar) using n-hexane-CH<sub>2</sub>Cl<sub>2</sub> gradient (increment 10%) up to 100% CH<sub>2</sub>Cl<sub>2</sub> and elution concluded with ethyl acetate, collecting 20 mL each. The process afforded various sub-fractions (**I-VI**) as determined by TLC [solvent systems: n-hexane-CH<sub>2</sub>Cl<sub>2</sub> (1:2, 2:3) and CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 99:1)]. The sub-fraction **I** (fractions 1-8) did not show any detectable spot and was not followed further. Sub-fraction **II** (fractions 9-15) showed a single spot R<sub>f</sub> 0.60 (eluent: n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 1:1), which upon crystallization in n-hexane-CH<sub>2</sub>Cl<sub>2</sub> mixture gave **143** (23 mg). Sub-fraction **III** (fractions (20-60) also showed a single spot R<sub>f</sub> 0.43 (eluent: n-hexane −CH<sub>2</sub>Cl<sub>2</sub>, 1:1), that was Dragendorff's reagent positive was subjected to repeated column chromatographic separation followed by evaporation of the solvent and afforded compound **144** (R<sub>f</sub> 45 mg). Sub-fraction **IV** (fractions 63-86, 3.7 g) showed two spots of R<sub>f</sub> 0.30 and 0.38 (eluent: n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 1:1) and upon repeated chromatography gave a further **144** (15 mg) and **145** (20 mg). Sub-fraction **V** (fractions 93-130, 3.4 g) showed one major spot of R<sub>f</sub> 0.40 and upon repeated chromatography gave **146** (165 mg). Fractions 134-180

constituted sub-fraction **VI** (5 g) was further purified by medium pressure chromatography (2.5 x 50 cm, 150 g, pressure  $\approx$  1 bar) to give a further amounts of compound **145** (50 mg) and **146** (70 mg) (eluent: *n*-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 1:1).

## 3.4.4 Physical and Spectroscopic data of compounds from CH<sub>2</sub>Cl<sub>2</sub> extract

## **3.4.4.1** Compound 143

Yellow powder  $R_f$  0.60 [n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, (1:1)], m.p. 255–257°C (Lit. 260-262°C; Wen et al., 2007); <sup>1</sup>H NMR and <sup>13</sup>C NMR data: (See **Table 20**, **page 98**, **Appendix 14**); **ESI-MS**: m/z (rel int.) 449 [ $M^+$ +1] (5), 432 (21), 431 (100), 414 (34), 384 (30), 354 (10), 330 (20).

## 3.4.4.2 Compound 144

Colourless powder  $R_f$  0.43 60 [n-hexane- $CH_2Cl_2$ , (1:1)], m.p. 92-94°C (Lit. 88-90°C; Abu Bakar  $et\ al.$ , 2007; Mohammad  $et\ al.$ , 2013); <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 21** and **22** on **page 101**, **102**, **Appendix 15**); **ESI-MS**: m/z (rel int.): 332 [M]  $^+$  (100), 331 (10), 276 (7), 250 (30), 248 (10), 210 (10).

## **3.4.4.3 Compound 145**

Colourless crystals, R<sub>f</sub> 0.38 60 [*n*-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 1:1)], m.p. 196-197°C (Lit. 194-195°C; Nayak *et al.*, 2010); <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 21** and **22**, **page 101**, **102**, **Appendix 15**); **ESI-MS**: *m/z* (rel int.): 294 [M] <sup>+</sup> (100), 296 (10), 293 (24).

## **3.4.4.4 Compound 146**

Pale yellow crystals R<sub>f</sub>, 0.40, m.p. 225-226°C (Lit. 224-225°C; Mohammad *et al.*, 2013); <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 21** and **22**, **page 101**, **102**, **Appendix 15**); **ESI-MS**: *m/z* (rel int.) 324 [M] <sup>+</sup> (100), 323 (50), 308 (5), 282 (5).

#### 3.4.5 Fractionation of MeOH extract

Medium pressure chromatographic separation of the MeOH extract (150 g) over 2% oxalic acid solution-deactivated silica gel column using a mixture of CH<sub>2</sub>Cl<sub>2</sub>-MeOH (5% -100% MeOH) gave a total of 150 fractions of 50 mL each. Fractions exhibiting similar TLC profiles were pooled together (Pools **I-III**). Pool **I** (fractions 7-20, 5 g) showed one major spot R<sub>f</sub> 0.50 (eluant: CH<sub>2</sub>Cl<sub>2</sub>-MeOH, 97:3) and was further purified by crystallization gave **65** (180 mg). Fractions 25-45 (pool **II**, 7 g) was repeatedly fractionated over 2% oxalic acid-deactivated silica gel (SiO<sub>2</sub> 150 g; 3.5 x 50 cm; 2-3 % MeOH-CH<sub>2</sub>Cl<sub>2</sub>) affording sub-fractions which resulted in the isolation of further amounts of **65** (15 mg), **68** (35 mg), **139** (70 mg). Pool **III** (5 g) gave **136** (35 mg) under similar purification procedure.

## 3.4.6 Acid hydrolysis of compounds 136 and 139

Compounds **136** and **139** (each 10 mg) in a mixture of 8% HCl (2 mL) and MeOH (20 mL) were separately heated under reflux for 2 h. The reaction mixtures were dried under reduced pressure to, dissolved in H<sub>2</sub>O (3 mL) and neutralized with NaOH. The neutralized products were then subjected to TLC (eluent: EtOAc–MeOH–H<sub>2</sub>O–HOAc, 6:2:1:1) and paper chromatography (PC) (eluent: n-BuOH–HOAc–H<sub>2</sub>O, 4:1:5). The sugar chromatograms were sprayed with aniline hydrogen phthalate followed by heating at 100 °C and the spots were identified after comparison with authentic samples. Similarly, the aglycones were confirmed upon exposure to conc. ammonia solution (Maciej, 2000).

## 3.5 Extraction of *Erythrina abyssinica* leaves

The air dried and pulverized leaves ( $\approx 2$  Kg) was soaked sequentially in *n*-hexane (3 x 3L), CH<sub>2</sub>Cl<sub>2</sub> (3 x 3L) and MeOH (3 x 3L), each lasting four days at room temperature with occasional

shaking using orbital shaker. The extracts were filtered and the filtrate evaporated to give 16 g, 27g and 86 g of dark green materials, respectively.

#### 3.5.1 TLC analysis of n-hexane and $CH_2Cl_2$ extracts

Thin layer chromatographic analysis of n-hexane extract (eluent: n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 2:3) showed a spot of R<sub>f</sub> 0.82 which turned bluish-purple with anisaldehyde spraying reagent after heating at 100 °C for 1 minute. On the other hand, TLC analysis of CH<sub>2</sub>Cl<sub>2</sub> extracts using n-hexane-CH<sub>2</sub>Cl<sub>2</sub> (2:3) followed using same solvent in the ratio 1:3 showed spots of R<sub>f</sub> 0.82, 0.69 and 0.52 respectively. The spot with R<sub>f</sub> 0.82 gave bluish-purple colour on spraying with anisaldehyde reagent followed by heating at 100 °C for a minute. The spot with R<sub>f</sub> 0.52 gave an orange coloration on spraying with the same reagent.

#### 3.5.2 Chromatographic separation of the *n*-hexane extract

Approximately 10 g of the n-hexane extract was dissolved in  $CH_2Cl_2$ , adsorbed on an equivalent amount of silica gel in  $CH_2Cl_2$  and then dried under vacuo. The free flowing material was then subjected to column chromatography (3 cm diameter by 60 cm long) glass column packed with silica gel ( $\approx 150$  g) using n-hexane, n-hexane- $CH_2Cl_2$  mixtures and finally with  $CH_2Cl_2$  as the elution solvent. A total of 130 fractions, each 20 mL were collected and their composition monitored using analytical TLC with solvent system n-hexane-  $CH_2Cl_2$  (4:1, 2:3). Fractions with similar profiles were combined into three pools (I-III). Pool I (fractions 1-30) did not show any spot on TLC and was not followed further. Fractions 33-40 were combined to give pool II which gave an orange oily substance which faded in colour after some time and was not followed further. Pool III (fractions 50-120, 1.5 g) showed a single spot of  $R_f$  0.82 (eluent- n-hexane-  $CH_2Cl_2$ , 2:3) which upon crystallization in  $CH_2Cl_2$ -MeOH mixture gave compound 133 (70 mg).

#### 3.5.3 Chromatographic separation of the CH<sub>2</sub>Cl<sub>2</sub> of extract

A portion of the  $CH_2Cl_2$  (25 g) was adsorbed onto silica gel and then subjected to column chromatography (3.0 x 60 cm, silica gel 200 g, and pressure $\approx$ 1 bar) using n-hexane- $CH_2Cl_2$  gradient (increment 10%) up to 100%  $CH_2Cl_2$  and finally eluted with 100% ethyl acetate. This process afforded sub-fractions (**I-IV**) as determined by the TLC profiles [solvent system (n-hexane- $CH_2Cl_2$ , 4:1, 2:3) and  $CH_2Cl_2$ -MeOH (99:1]. Sub-fraction **I** (Fractions 1-30) did not show any spot and was not followed further. Sub-fraction **II** (Fractions 35-90, 1.5 g) showed a single spot of  $R_f$  value 0.63 (solvent: n-hexane-  $CH_2Cl_2$  (solvent: n-hexane-  $CH_2Cl_2$ , 2:3) and upon evaporation of solvent followed by crystallization in n-hexane-  $CH_2Cl_2$  mixture afforded **134** (50 mg). Sub-fraction **III** (Fractions 92-104, 2.5 g) showed a single spot of  $R_f$  0.52 (eluent: n-hexane- $CH_2Cl_2$ , 2:3) which upon crystallization in n-hexane- $CH_2Cl_2$ , gave compound **147** (55 mg). Sub-fraction **IV** (Fractions 140-186, 5 g) showed two spots of  $R_f$  values 0.52 and 0.30 (eluent: n-hexane- $CH_2Cl_2$ , 2:3) which on repeated chromatography afforded a further **147** in 15 mg and compound **148** (45 mg).

#### 3.5.4 Fractionation of the MeOH extract

A portion of the extract (40 g) was adsorbed on silica gel and then subjected to column chromatography with CH<sub>2</sub>Cl<sub>2</sub> containing increasing amounts of MeOH (gradient elution with increasing concentration of the polar solvent (increment 3%) and elution concluded with 100% MeOH. A total of 50 fractions were sampled and their composition analysed by TLC (eluent: CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 98:2, 97:3, 19:1, 9:1 and 4:1) and those exhibiting similar TLC profiles were combined. This process afforded two sub-fractions (I and II) as determined by the TLC profiles [solvent system (*n*-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 3:1, 1:2) and CH<sub>2</sub>Cl<sub>2</sub>–MeOH (97:3]. Sub-fraction I (Fractions 10-26, 3 g) contained one major spot of R<sub>f</sub> value 0.30 (eluent: *n*-hexane-CH<sub>2</sub>Cl<sub>2</sub>, 2:3)

and was similarly fractionated as in **3.5.3** above to give compound **147** (15 mg). Pool **II** (Fractions 32-105, 6 g) was similarly resolved into individual components using CH<sub>2</sub>Cl<sub>2</sub>-MeOH (98:2) followed by the same solvent in the ratio 98:3 to give **148** (21 mg).

## 3.5.5 Physical and spectroscopic data of compounds isolated from *Erythrina abyssinica*

## 3.5.5.1 Compound 147

Grenish-yellow powder,  $R_f$ , 0.30, [n-hexane-CH<sub>2</sub>Cl<sub>2</sub>, (2:3)], m.p. 194-196°C (Lit. 192-193°C; Yenesew et~al., 2009);  ${}^{1}$ H and  ${}^{13}$ C NMR data: (See **Table 23, page 105**, **Appendix 16**); **LCMS**: m/z (rel int.) 336 [M]  ${}^{+}$  (80), 321 (21), 279 (25),149 (100), 137(79), 115 (25), 108 (20), 69 (15), 55 (14), 41 (25).

## 3.5.5.2 Compound 148

White powder, R<sub>f</sub> 0.45 [CH<sub>2</sub>Cl<sub>2</sub>-MeOH (98:2)], m.p. 112-114°C (Lit. 114-116°C; Nkengfack, *et al.*, 1997; Yenesew *et al.*, 1997); <sup>1</sup>H and <sup>13</sup>C NMR data: (See **Table 24, page 107**, **Appendix 17**); **LCMS**: *m/z* (ret int.): 588 [M] <sup>+</sup> (100), 573 (30), 560 (40), 559 (98), 554 (15), 561 (10).

## 3.6 Biological activity studies

## 3.6.1 *In vitro* antiplasmodial assay

An *in vitro* antiplasmodial activity was carried out using the *P. falciparum* multi-drug resistant Indochicha I (W2) and chloroquine-sensitive Sierra Leone I (D6) strains according to procedures of Desjardins *et al.*, (1979) and Chulay *et al.*, (1983) to determine the IC<sub>50</sub>. The parasites were grown in a continuous culture supplemented with mixed gas (90% nitrogen, 5% oxygen), 10% human serum and 6% haematocrit a, plus red blood cell (Trager and Jensen, 1976). When the cultures had reached a parasitemia of 3% with at least a 70% ring developmental stage present, parasites were transferred to a 96 well microtitre plate precoated with the extracts/isolates dissolved in DMSO. The samples were diluted across the plate to provide a range of concentrations to determine IC<sub>50</sub> values. The plate was then incubated in a mixed gas incubator for 24 h. Thereafter 3H-hypoxanthine was added and the parasite allowed to grow for 18 h. Cells were then processed with a plate harvester (TomTec) on a filter paper and washed to eliminate unincorporated radioisotope. Mefloquine (1  $\mu$ g/mL) was used as a standard drug. The experiments were done in triplicate-(MEMRI-Kenya).

### 3.6.2 Larvicidal and mosquitocidal tests

1 mg of crude extracts and isolates were weighed and dissolved in 1 mL acetone (analytical grade, Lobarchemi) to give a stock solution of 1000  $\mu$ g/mL. The larvae were exposed to a wide range of test concentrations and a control to find out the activity range of the crude samples and isolates. After determining the mortality of larvae in this wide range of concentrations, a narrow range of five concentrations (10, 100, 250, 500  $\mu$ g/mL) yielding up to 100% mortality in 24 h was used to determine LC<sub>50</sub> values. Temperatures were maintained at 26±3°C while humidity

was kept at 70-75%. Batches of 20 3<sup>rd</sup> instar Anopheles gambie larvae were transferred by means of droppers into a small disposable test cup each containing 50 mL of the extract and isolates at concentrations of 10, 100, 250, 500 µg/mL dissolved in acetone and fed on larval food (Globade et al., 2002). Temephos (3 µg/mL) was used as positive control, while 1% acetone was used as negative control. After 24 hours, larval mortality was recorded whereby moribund larvae were counted and added to the dead ones for calculating % mortality. Dead larvae were considered as those which could not be induced to move when they were probed with a needle in the siphon region. For mosquitocidal assays, sample solutions of 10, 100, 250, 500 µg/mL dissolved in acetone were impregnated on Whatman filter paper No. 1 measuring 140 x 100 mm. The bioassay was conducted in an experimental kit consisting of two cylindrical plastic tubes, both measuring 125 x 44 mm (Globade, et al., 2002). One tube served to expose the mosquitoes to the extracts and isolates while another tube was used to hold the mosquitoes before and after the exposure periods. The impregnated papers were rolled and placed in the exposure tube. Each tube was closed at one end with a 16 mesh size wire screen. Sucrose-fed and blood starved female mosquitoes (20) were released into the tube, and the mortality effects of the extracts and isolates were observed every 10 minutes for 3 hours exposure period. At the end of 1, 2, and 3 hrs exposure periods, the mosquitoes were placed in the holding tube. Cotton pads soaked in 10% sugar solution with vitamin-B complex were placed in the tube during the holding period of 24 hours and mortality rate was recorded (Globade, et al., 2002). Any mosquito was considered to be dead if did not move when prodded repeatedly with a soft brush. The above procedure was carried out in triplicate for each concentration. The mortality of the mosquitoes was monitored and toxicity levels of the test samples evaluated graphically to give LC50 values.

Lambdacyhalothrin (1  $\mu$ g/mL) was used as a positive control while 1% acetone was employed as a negative control. The tests were done in triplicate.

#### 3.6.3 Antifungal and antibacterial tests

The disc diffusion assay method was applied (Singh et al., 2002) using clinical isolates of Candida albicans (HG 392), Aspergillus fumigatus (HG 420) and Aspergillus niger (ATCC 90028) as the representative fungi. Staphylococcus aureus (ATCC 25922), Streptococcus faecalis (ATCC 25925) and Bacillus anthracis (QST 713) were used as the representative Gram positive bacteria while Klebsiela pneumaoniaie (ATCC 90028), Salmonella typhimurium (ATCC 25927), Pseudomonas aeruginisa (ATCC 25923) and E. coli (K 12) were representatives of Gram negative bacteria. Crude samples were tested in vitro at sample concentration of 1000 μg/mL dissolved in dimethylsulfoxide (DMSO), while pure isolates were tested at 100 μg/mL. Mueller Hinton agar was aseptically aliquoted at volumes of 25 mL into petri dishes and left to congeal. 250 µL of fungi/bacteria cell suspension (at a concentration of 10<sup>8</sup> CFU/mL) was spread onto the surface of agar medic in petri dish (Singh et al., 2002). Sterile paper discs (Schleicher and Schuell type 602 H, Germany, 5 mm diameter) previously impregnated with 10 µL of test samples were placed approximately equidistant into the seeded agar using a sterile forceps. This was done in triplicate. Disc containing 10µL of the 20 µg/mL solution of the standard drug (fluconazole and amoxyllin, respectively) was used as the positive control. The agar plates were incubated at 37°C for 24 h after which the inhibition zones were measured in millimeters (McChesney et al., 1991).

## 3.7 Data analysis

#### 3.7.1 Antiplasmodial activity

The filters were measured for activity in a microtitre plate scintillation counter (Wallace). Data from the counter were imported into a Microsoft Excel Spreadsheet, which was then imported into an Oracle Database Program to determine  $IC_{50}$  values (Trager and Jensen 1976).

## 3.7.2 Larvicidal and mosquitocidal Activity

The activity of the extracts and the isolates were evaluated according % mortality scale; (>75%: highly active; 50-74%: moderate; 25-49%: weak; <25%: inactive (Globade, *et al.*, 2002). The results are also reported as lethal concentration;  $LC_{50}$  for extracts and each compound and the minimum concentration that gives 50% larvicidal activity was determined in  $\mu$ g/mL. Average 24 hour knock down (or mortality) data were compared with both negative and positive control (Globade *et al.*, 2002).

#### 3.7.3 Antifungal and antibacterial activity

The results are reported as zones of inhibition measured in mm. The size of the zone of inhibited growth (in mm) indicated the degree of antimicrobial susceptibility according to the scale Activity scale: (> 17: Highly active; 11-16: intermediate; 7-10: weak; <6: resistant) (McChesney et al. 1991).